

## Properties

### Resistance to chemicals of expanded materials

Styropor foam's resistance to chemicals corresponds to that of parts made from polystyrene. However, because Styropor's cell structure gives the material a greater surface area, damage occurs quicker, and to a greater extent, than is the case with the compact basic polystyrene material. Similarly, low density Styropor is attacked far more easily than higher density Styropor.

In practice (eg, in the construction or packaging sectors) it is very important to know how Styropor reacts to chemical substances in order to prevent defects.

#### Test

The test is based on DIN 53428 "Testing of cellular materials; Determination of the resistance to liquids, vapours, gases and solid materials". In this DIN standard, 5 foam cubes (skinless) of side 5 cm are immersed in the test medium for a definite length of time and changes to mass and size measured. The immersion time depends on the test medium: for liquids it is 72 hours; for gases 24 hours; and for liquefied gases, at least three hours.

For liquefied gases the immersion temperature is at, or just under, the boiling point of the test medium; in other media, immersion takes place at room temperature.

For visual assessment of damage, DIN 53428 suggests a scale of criteria from 0 (no change) to 5 (severely damaged). To give a simpler overview, the table overleaf contains the following assessment criteria:

- + = unchanged ( $\Delta$  0)  
= resistant
- +– = slight change ( $\Delta$  2)  
= limited resistance  
(small change in size)
- = severely damaged ( $\Delta$  5)  
= not resistant

If Styropor foam comes into contact with unknown substances that could contain damaging solvents (eg, paint or adhesive) one can start by making sure the foam is not attacked by carrying out a trial under practical conditions. The trial may be shortened considerably if it is carried out at temperatures above 20 °C (eg, 50 °C). To get clearer evidence of the foam's resistance, the severity of the test conditions can be increased by testing a foam whose density is much lower than that intended for the actual application.

The table overleaf shows the resistance of Styropor foam to the most common chemical substances.

Substance	Styropor P & F foams	Substance	Styropor P & F foams	Substance	Styropor P & F foams
Sea water	+	<b>Liquefied gases:</b>		<b>Inorganic building materials:</b>	
Water	+	a) inorganic		Anhydrite	+
<b>Alkalis:</b>		Ammonia	+	Cement	+
Ammonia water	+	Hydrogen	+	Chalk	+
Bleaching agents (hypochlorite, hydrogen peroxide)	+	Inert gases	+	Gypsum	+
Caustic soda solution	+	Nitrogen	+	Sand	+
Lime water	+	Oxygen	+	<b>Organic building materials:</b>	
Potassium hydroxide solution	+	(risk of explosion)		Bitumen	+
Soap solutions	+	Sulfur dioxide	-	Cold asphalt and solvent-based bituminous cement (free from aromatics)	-
<b>Diluted acids:</b>		b) organic		Cold asphalt and water-based bituminous cement	+
Acetic acid, 50 %	+	Butadiene	-	<b>Aromatics:</b>	
Formic acid, 50 %	+	Butane	-	Benzene	-
Hydrochloric acid, 18 %	+	Ethene	+	Cumol	-
Hydrochloric acid, 7 %	+	Ethene oxide	-	Ethyl benzene	-
Hydrofluoric acid, 4 %	+	Ethyne (acetylene)	-	Phenol, aqu. soln. 1 %	+
Hydrofluoric acid, 40 %	+	Methane	+	Phenol, aqu. soln. 33 %	-
Nitric acid, 13 %	+	Natural gas	+	Styrene	-
Nitric acid, 50 %	+	Propane	-	Toluene	-
Phosphoric acid, 50 %	+	Propene	-	Xylene	-
Phosphoric acid, 7 %	+	Propene oxide	-	<b>Vapours of:</b>	
Sulfuric acid, 10 %	+	<b>Aliphatic hydrocarbons:</b>		Camphor	-
Sulfuric acid, 50 %	+	Cyclohexane	-	Naphthalene	-
<b>Concentrated acids:</b>		Diesel fuel, Heating oil	-	Styropor FH can be used to produce foams that have increased resistance to aromatic-free hydrocarbons compared with other Styropor grades. The suitability of this product for a particular application must be tested in each individual case.	
Acetic acid, 96 %	-	Gasoline	-	<b>Note</b>	
Formic acid, 99 %	+	(regular & super grade)		The information submitted in this publication is based on our current knowledge and experience. In view of the many factors that may affect processing and application, these data do not relieve processors of the responsibility of carrying out their own tests and experiments; neither do they imply any legally binding assurance of certain properties or of suitability for a specific purpose. It is the responsibility of those to whom we supply our products to ensure that any proprietary rights and existing laws and legislation are observed.	
Hydrochloric acid, 36 %	+	Heptane	-		
Nitric acid, 65 %	+	Hexane	-		
Propanoic acid, 99 %	-	Paraffin oil	+/-		
Sulfuric acid, 98 %	+	Vaseline	+		
<b>Fuming acids:</b>		White spirit 155–185 °C	-		
Nitric acid	-	White spirit 55–95 °C	-		
Sulfuric acid	-	<b>Alcohols:</b>			
<b>Anhydrides:</b>		Butanol	+/-		
Carbon dioxide, solid	+	Coconut oil alcohol	+		
Ethanoic anhydride (acetic anhydride)	-	Cyclohexanol	+		
Sulfur trioxide	-	Diethylene glycol	+		
<b>Weak acids:</b>		Ethanol	+/-		
Carbonic acid	+	Ethylene glycol	+		
Citric acid	+	Glycerin	+		
Humic acid	+	Iso-propanol	+		
Lactic acid	+	Methanol	+/-		
Tartaric acid	+	<b>Amines:</b>			
<b>Gases:</b>		Aniline	-		
a) inorganic		Diethyl amine	-		
Ammonia	-	Ethyl amine	+		
Bromine	-	Triethyl amine	-		
Chlorine	-	<b>Miscellaneous organic substances:</b>			
Sulfur dioxide	-	Acetone	-		
b) organic		Acetone nitrile	-		
Butadiene	-	Acrylonitrile	-		
Butane	-	Dimethylformamide	-		
Butene	-	Esters	-		
Ethane	+	Ethers	-		
Ethene (ethylene)	+	Halogenated hydrocarbons	-		
Ethyne (acetylene)	+	Ketones	-		
Methane	+	Olive oil	+		
Natural gas	+	Paint thinners	-		
Propane	+	Tetrahydrofuran	-		
Propene (propylene)	+				
Propene oxide	-				

BASF Aktiengesellschaft  
D-67056 Ludwigshafen, Germany

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